Effects of manganese additive on piezoelectric properties of (Bi_{1/2}Na_{1/2})TiO₃-BaTiO₃ ferroelectric ceramics

GuiFen Fan · WenZhong Lu · XiaoHong Wang · Fei Liang

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Abstract The microstructure, piezoelectric and dielectric properties of piezoelectric ceramics (Bi_{1/} $_2Na_{1/2})_{0.94}Ba_{0.06}TiO_3 + xwt\%MnCO_3$ (BNBT6-xMn)with x ranging from 0 to 1.5 were studied. X-ray diffraction patterns showed the samples exhibited perovskite structure and Mn additive turned the structure from tetragonal phase to rhombohedral phase. The results of SEM indicated the Mn additive promoted the sintering which was beneficial to the synthesis of the perovskite. With the amount of Mn additive up to 0.1 wt.%, the value of d_{33} , k_p , and ε were improved; While $Q_{\rm m}$ also was increased until 0.8 wt%. The Mn additive acted on the "soft" and "hard" function simultaneously; When the amount of Mn additive more than 0.8 wt%, the performance drops.

Introduction

Bismuth sodium titanate, (Bi_{1/2}Na_{1/2})TiO₃ (BNT), is the A-site compound perovskite ferroelectric body, discovered in 1960 by Smolenskii [1]. At room temperature, the BNT is a rhombohedra ferroelectrics (*a*=0.3886 nm, α =89.6°) which has a high Curie temperature, *T*_c=320 °C and a phase transition point from ferroelectric to anti-ferroelectric, *T*_p=230 °C. The BNT shows strong ferroelectric properties because of relatively large remanent polarization, $P_r=38 \ \mu\text{C/cm}^2$, and a large coercive field, $E_c=73 \text{ kv/cm}$. So it is a kind of very promising lead-free piezoelectric ceramic material for environmental protection. However, the BNT is hardly applied to practice because of the high coercive field and high conductivity that make it difficult to be poled.

In order to resolve these problems, some modifications on BNT composition have been studied. It has been reported that BNT-based composition is modified with SrTiO₃ [2], CaTiO₃ [3], BaTiO3 [4], NaNbO₃ [5], BiFeO₃ [6] ,and MnCO₃, Fe₂O₃, NiO [7] .Among these BNT-based systems, $(1-x)(Bi_{1/2}Na_{1/2})$ TiO₃-xBaTiO₃ is more interesting. Its morphotropic phase boundary (MPB) composition is at x = 0.06-0.07, where the ceramic presents relatively high piezoelectric properties. In this paper, for further enhancement on the piezoelectric properties of 0.94(Bi_{1/2}Na_{1/2})TiO₃-0.06BaTiO₃(BNBT6), the microstructure, piezoelectric and dielectric properties of Mn-doped BNBT6 ceramic were studied as a function of the amount of MnCO₃. The modification mechanism of added MnCO₃ was also discussed.

Experimental

The conventional ceramic fabrication technique was used to prepare $(Bi_{1/2}Na_{1/2})_{0.94}Ba_{0.06}TiO_3+xwt\%Mn-CO_3$ (BNBT6-*x*Mn) (*x*=0, 0.05, 0.1, 0.3, 0.5, 0.8, 1, 1.5) ceramics. Reagent-grade metal oxide or carbonate powders of Bi₂O₃, Na₂CO₃, BaCO₃, and TiO₂ were used as the starting materials. The powders of these raw materials were mixed in alcohol with zirconium balls by ball-milling for 10 h and calcined at 880°C for

G. Fan (⊠) · W. Lu · X. Wang · F. Liang Department of Electronic Science and Technology, HuaZhong University of Science and Technology, WuHan 430074, P.R. China e-mail: hustfgf2009@163.com

2 h. The MnCO₃ at different amount up to 1.5 wt% was added into the calcined powders. Then the mixture was milled in alcohol for 10 h, dried and granulated with polyvinyl alcohol (PVA) as a binder. The granulated powder was pressed into discs in diameter of 20 mm and thickness of 1.5 mm for samples, and these discs were sintered at 1160 °C for 2 h in an air atmosphere. The crystal phase of sintered ceramics was checked using an X-ray diffractometer (Rigaku Dmax-Rc, Japan). Microstructures of these samples were observed with a field emission scanning electron microscope (FESEM) (FEI-Sirion2000, Phillip, Holland). The composition analysis of the sample was measured using an EMPA Analyzer (JXA-8800R, Japan). The density was measured by Archimedes' method.

Fired-on silver paste was used as the electrode for dielectric and piezoelectric measurements. The specimens for measurement of piezoelectric properties were poled in silicon oil at R.T~60 °C with DC electric field of 3.5-4 kv/mm for 10 min. The piezoelectric electric charge coefficient (d_{33}) of the samples were measured using a piezo-meter(ZJ-2,) at a frequency of 100 Hz. The dielectric constant, ε , and loss tangent, tan δ , were measured with a capacitance- meter(TH2613). Piezoelectric properties were measured by means of a resonance-antiresonance method on the basis of IEEE standards using an impedance analyzer (Agilent 4192A). The electric-mechanical coupling factor, $k_{\rm p}$, and mechanical quality factor, $Q_{\rm m}$, were calculated from the resonance and antiresonance frequencies. The P-E hysteresis loops of the samples were measured using an Analyzer (ZT-1).

Results and discussion

Figure 1 shows X-ray diffraction patterns of BNBT6– xMn ceramics for x=0, 0.1, 0.3, 0.8 and 1.5. The pattern of BNBT6 ceramic indicates that its crystal structure is the perovskite structure with the coexistence of rhombohedral and tetragonal except small Na (Ti_2O_4) phase. With the increase of the amount of Mn additive, the Na (Ti_2O_4) phase gradually becomes smaller. With the amount of Mn additive up to 1.5 wt%, the crystal structure is pure single perovskite and no Na (Ti_2O_4) phase.

Figure 2 displays the SEM micrographs of BNBT6–xMn samples. It is obvious that the Mn addition in BNBT6 ceramics leads to the increase of the average gain size and the change of grain shape from oval to cubic at the same sintering temperature. In the case of Mn addition at amount less than 0.8 wt%, the grain



Fig. 1 The XRD patterns of BNBT6-xMn ceramics

size is uniform and the boundary of grains is very clear. But when the amount of Mn addition is up to 1.5wt%, the grain becomes too big and many air holes are observed, that makes the decrease of the density of samples. Figure 3 is the micrograph and electron spectrum of electron probe microanalysis (EMPA) of heat erosion BNBT6–1.5Mn sample. There are some educt at the grain boundary or defect when the sample is treated by heat erosion. Correspond to the composition analysis, the content of Mn addition in the educt or at the grain boundary is more than that in the grain. It seems that the Ti ions are ejected by Mn ions to the grain boundary, so that the amount of Ti ions has relatively high value. Therefore, when the content of Mn addition is up to 1.5wt%, partial Mn addition becomes precipitation phase and exists in the grain boundary or the crystals defect, which deteriorate the properties of the sample. Whereas, proper Mn addition may enhance the grain growth and promote the sintering.

Figure 4 shows the P–E hysteresis loops of BNBT6– xMn ceramics samples at room temperature. With increasing amount of Mn, the coercive field E_c increases, from 2.9 kv/mm for x=0, to 4.1 kv/mm for x=1.0.The tendency of E_c is different from the result of X.Y.Zhou's et.al research [8].

The density ρ and the piezoelectric charge coefficient d_{33} of BNBT6–*x*Mn ceramics as functions of MnCO₃ content are shown in Fig. 5. As shown in Fig. 5, the ρ and the d_{33} increase evidently and reach the maximum value which is 5.81 g/cm³ and 132pC/N, respectively, for the amount of 0.1wt% MnCO₃, then

Fig. 2 The SEM micrographs of BNBT6-xMn ceramics: (a) x=0; (b) x=0.3; (c) x=0.8 (**d**) *x*=1.5



Fig. 3 The micrograph and electron spectrum of electron probe microanalysis (EMPA) of heat erosion BNBT6-1.5Mn sample: (a) SEM micrograph; (b) SEM micrograph with different

locations (c) electron spectrum at +1 location;(c) electron spectrum at +2 location; (d) electron spectrum at +3 location

cps

250

200

150

100

50

0

O

Na

(c)



(a)

-6

(c)

- ρ

È

1.5

(b) x=0.3; **(c)** x=0.8; **(d)** x=1

140

120

100

80

60

40

0.0

d₃₃/(pC.N⁻¹



Fig. 5 The dependence of d_{33} and the density ρ dependence on the amount of Mn in BNBT-xMn ceramics

the amount of Mn/wt%

1.0

0.5

decrease rapidly. Figure 6 shows the mechanical quality factor $Q_{\rm m}$ and the planar electromechanical coupling factor k_p as function of MnCO₃ content. The maximum value of the $Q_{\rm m}$ is 417 at the amount of 0.8 wt% MnCO₃. The k_p has a maximum value of 32% at the amount of 0.1wt% MnCO₃, which is much larger than those of pure BNBT6 ceramics.

Figure 7 shows the dielectric constant, ε , and the loss tangent, tan δ , of BNBT6-xMn ceramics before or after poling. Before poling, the ε reduces with increasing MnCO₃ content; the tan δ is decreased and reaches the minimum at the amount of 0.8wt% and then increases shapely. After poling, the ε increases up to 0.5 wt%

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Fig. 6 The mechanical quality factor $Q_{\rm m}$ and the planar electromechanical coupling factor k_p dependences on the amount of Mn in BNBT-xMn ceramics

and reaches the maximum, the tan δ decreases up to 0.3wt% and then increases and especially increases shapely after 1 wt% MnCO₃. The ε and the tan δ before poling are almost larger than those after poling.

The k_p , d_{33} and ε_{33}^{T} of these samples increase with MnCO₃ content below 0.1 wt%. This is the so-called "soft" characteristics. The $Q_{\rm m}$ and the $E_{\rm c}$ of these samples increases with MnCO₃ content below 0.8wt%, which presents "hard" characteristic. This is to say, the Mn-doped BNBT6 ceramics show the properties of "soft" and "hard" piezoelectrics simultaneously below 0.8wt%MnCO₃. The results are agreement with the effect of Mn-doped PZT ceramics [9]. When MnCO₃

Fig. 7 The dielectric properties of BNBT6-xMn before or after poling: (a) Dielectric constant ε ; (b) Dielectric loss tan δ



content is over 0.8 wt%, the excess Mn ions and ejected Ti ions accumulate in the grain boundary to form the precipitated phase which makes the piezo-electric and dielectric properties of these samples deteriorated.

The mechanism of the effect of MnCO₃ presents two aspects: First, as MnCO₃ content increases, the tetragonality of BNBT6 ceramics and c/a ratio decrease, which facilitated the movement of domain wall to improve the properties of k_p , d_{33} , ε and P_r significantly. Secondly, it is known that MnO₂ can convert to other forms when heated [10]:

$$MnO_{2} \xrightarrow{535^{\circ}C} Mn_{2}O_{3} \xrightarrow{1080^{\circ}C} Mn_{3}O_{4} \xrightarrow{1650^{\circ}C} MnO$$

So in the course of sintering, Mn ions chiefly exist in the BNBT6 ceramics in two valence states: Mn²⁺ in radius of 0.067 nm and Mn³⁺ in radius of 0.058 nm. As a result of their radius close to the radius of Ti⁴⁺, Mn²⁺ and Mn³⁺ may substitute the Ti⁴⁺ of B-site. To compensate the charge, Mn ions function as an acceptor additive leading to bringing some oxygen vacancies in the crystal lattice which make oxygenoctahedron shrink and suppress the movement of domain wall. The result is the increase of $Q_{\rm m}$ and $E_{\rm c}$ and the decrease of $k_{\rm p}$. Generally, the two aspects of mechanism affect the properties of the material at the same time. Accordingly the Mn-doped BNBT6 ceramics show the properties of "soft" and "hard" piezoelectrics simultaneously below 0.8 wt% MnCO₃. Just, the replacement Mn ions for Ti ions which act as an acceptor additive affects the properties rather than the crystal structure change which is the decrease of the tetragonality and exhibit "soft" function.

Conclusion

The microstructure and piezoelectric properties of BNBT6–xMn ceramics have been investigated. X-ray diffraction patterns indicate the decrease of the tetragonality with the increase of Mn additive. The result of SEM shows that Mn additive makes the grains grow and promotes the sintering. The effect of Mn additive on the piezoelectric and dielectric properties of the BNBT6 ceramics are double—"Soft" and "hard" functions at the amount of Mn additive below 0.8wt%. At some extent, Mn additive may improve the properties of piezoelectric ceramics.

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